



Explore Broadband Near-Infrared Phosphor Perovskite $\text{LaLuO}_3:\text{Cr}^{3+}$

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INTRODUCTION

The NIR-phosphor as indispensable role decides the properties and function of pc-LEDs. Thus, it is important to explore suitable phosphors. The transition metal Cr^{3+} is an ideal NIR emitter and the luminescence of Cr^{3+} ions can be tuned via crystal field engineering easily. Accordingly, perovskite structure oxide with diversified structure and composition can provide the variety of local coordination environment for Cr^{3+} ions. LaLuO_3 with perovskite structure was chosen as host for the Cr^{3+} activators. A novel phosphor with broadband emission spectrum centered at 880 nm were found and has the application foreground.

METHODS

A series of $\text{LaLu}_{1-x}\text{O}_3:\text{xCr}^{3+}$ ($\text{LLO}:\text{xCr}^{3+}$, $0 \leq x \leq 0.06$) phosphors were successfully synthesized by solid-state reaction at high temperature. The stoichiometric raw materials La_2O_3 (99.99%), Lu_2O_3 (99.99%), Cr_2O_3 (99.95%) were homogeneously mixed in an agate mortar for more than 20 minutes. Then the powder blends were transferred into corundum crucibles and sintered at 1450 °C for 6 h. Finally, the synthesized samples were cooled to room temperature in a protective atmosphere and ground into powder for subsequent measurements.

RESULTS AND CONCLUSIONS

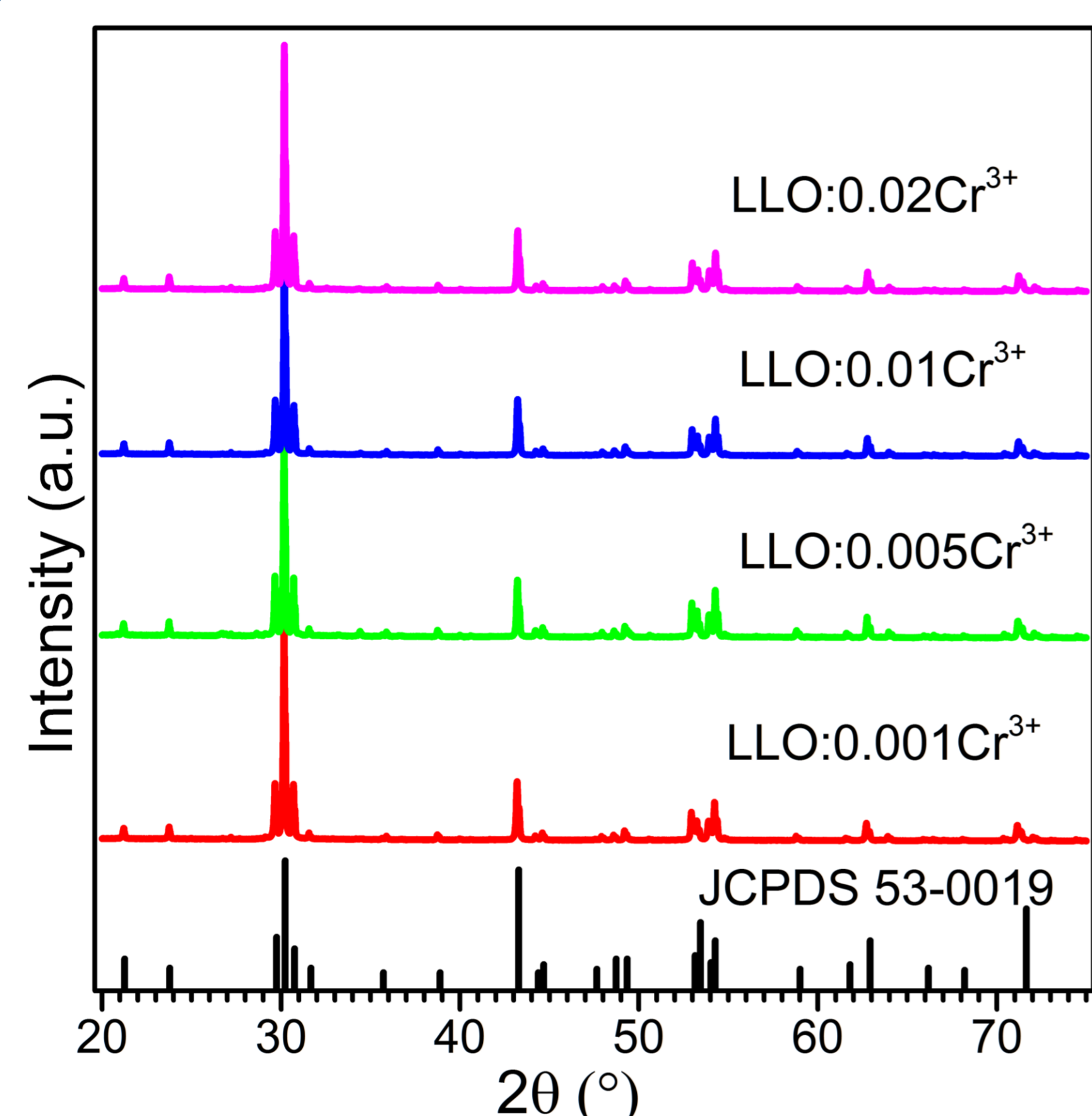


Fig.1. XRD patterns of $\text{LLO}:\text{xCr}^{3+}$ samples and standard data of LLO phase (JCPDS No. 53-0019)

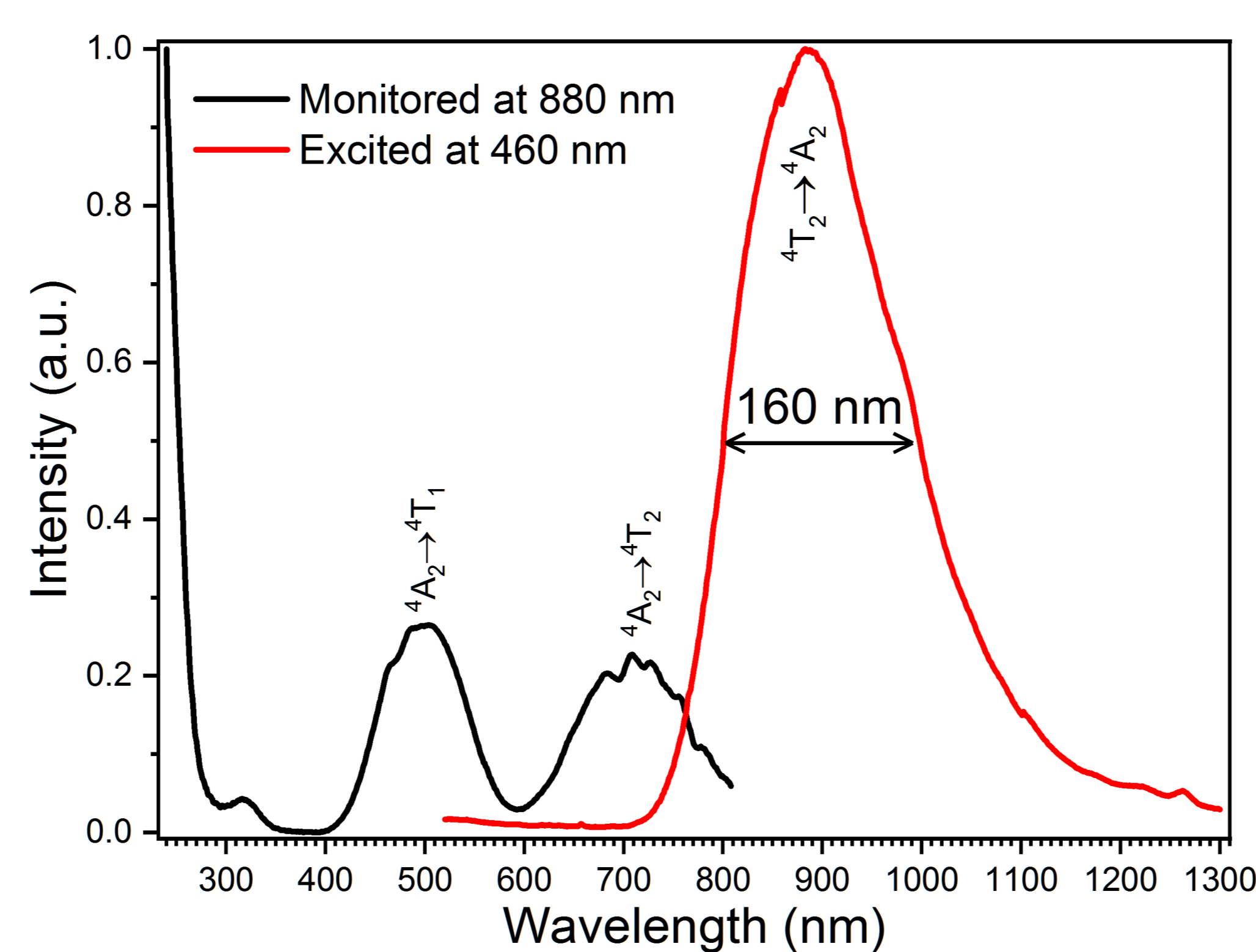


Fig.2. PL and PLE of $\text{LLO}:\text{1.5\%Cr}^{3+}$ at room temperature.

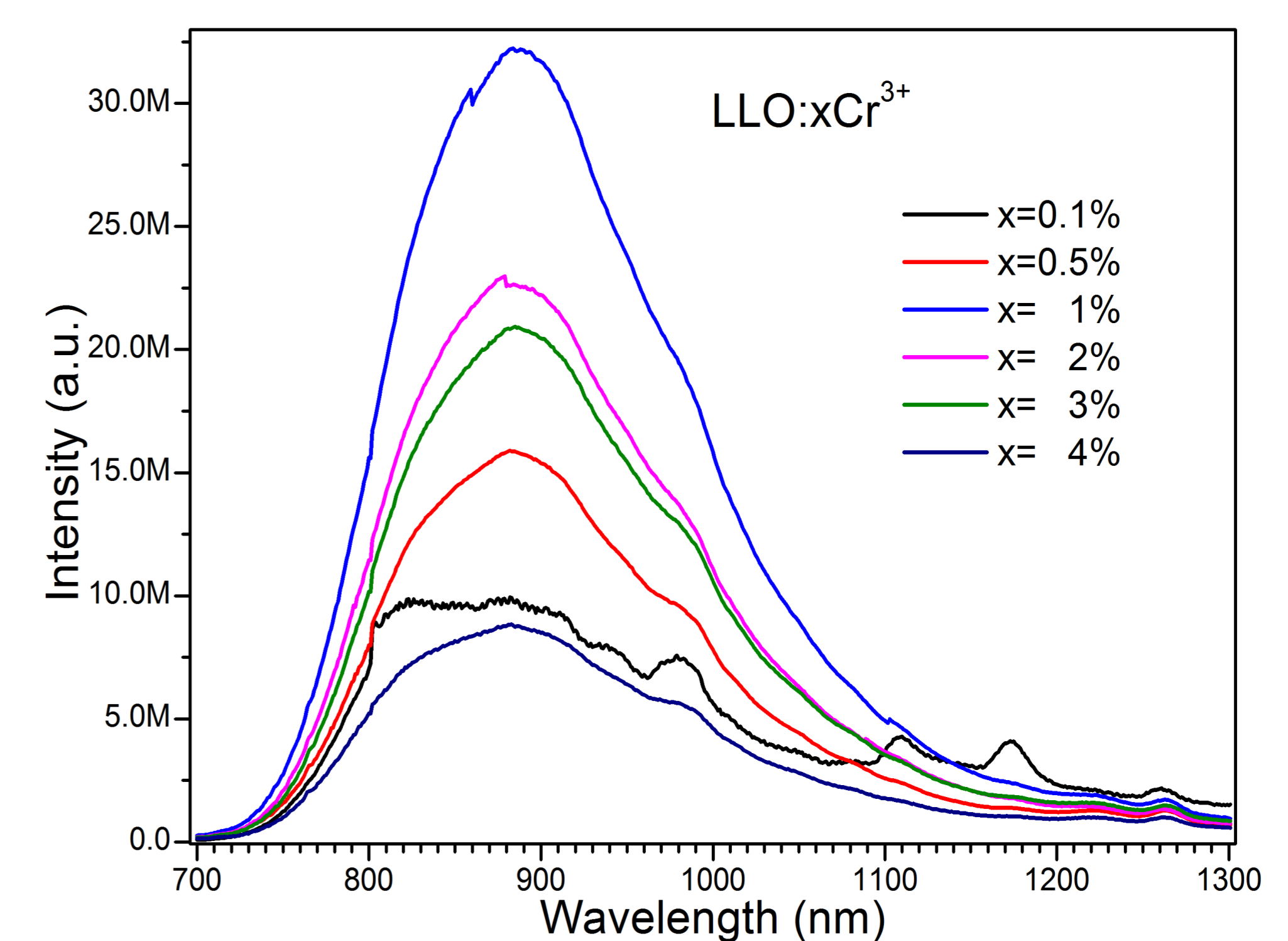


Fig.3. PL spectra of $\text{LLO}:\text{xCr}^{3+}$

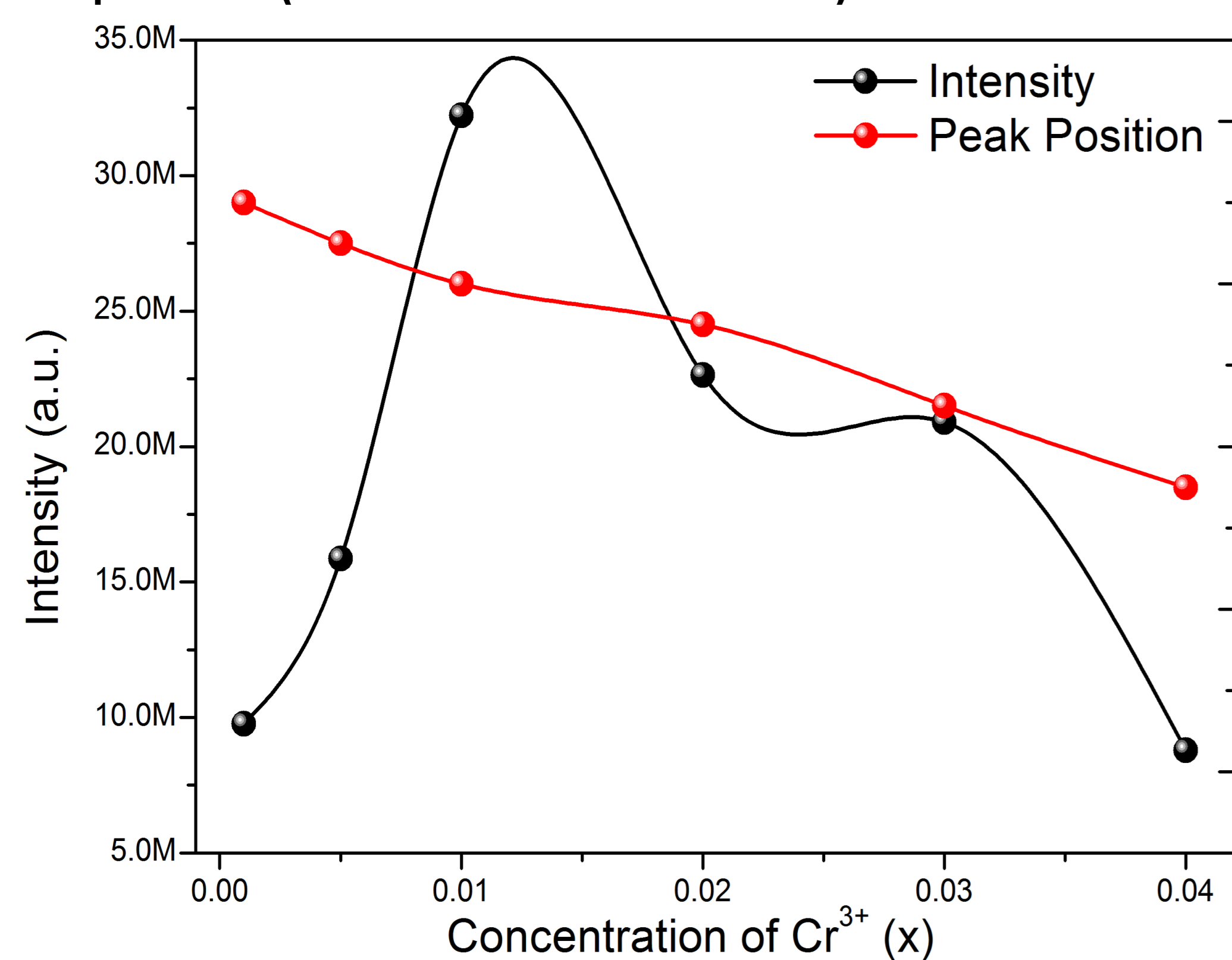


Fig.4. Peak position and Intensity changes with Cr^{3+} concentration

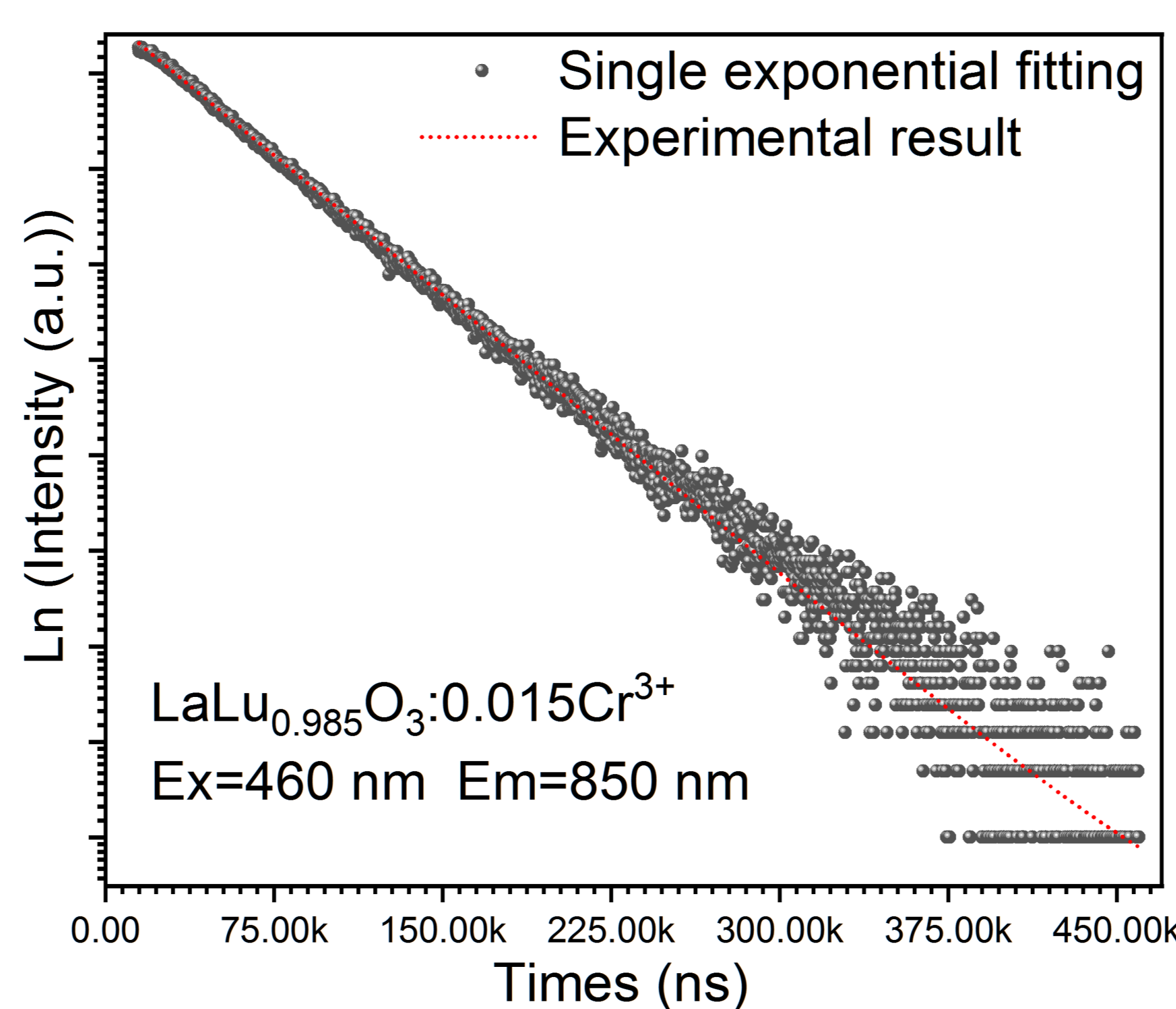


Fig.5. Decay curve of spectra of $\text{LLO}:\text{0.015Cr}^{3+}$ monitored at 850nm.

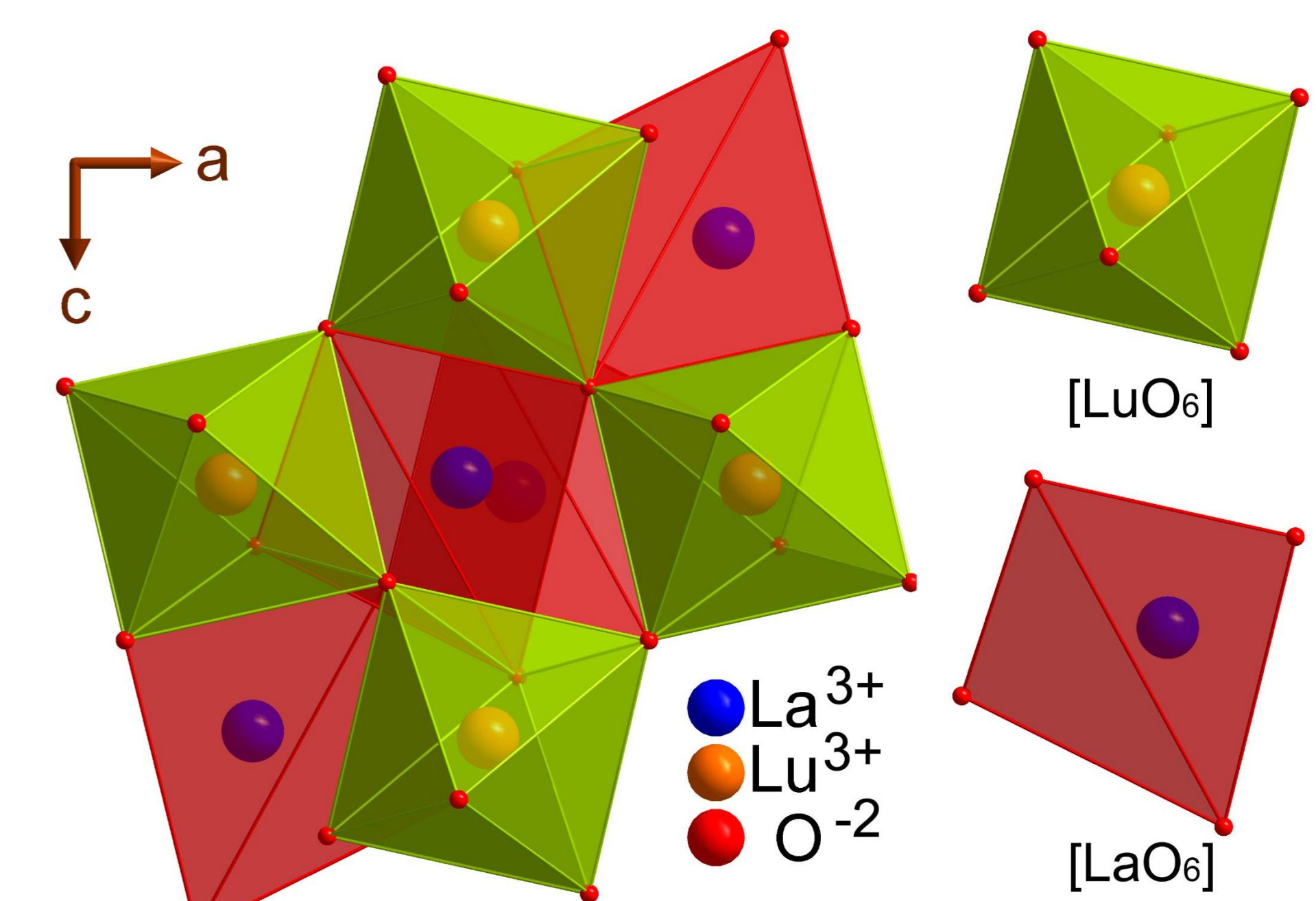


Fig.6. Crystal structure of LLO.

Conclusions

Cr^{3+} ions doped broadband near-infrared phosphor is obtained. PL is identified to originate from one available octahedral sites for Cr^{3+} ions. The optimal doping concentration of Cr^{3+} ions is 1.5%. The QY of $\text{LLO}:\text{0.01Cr}^{3+}$ might be 26.5%.